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## Key indicators

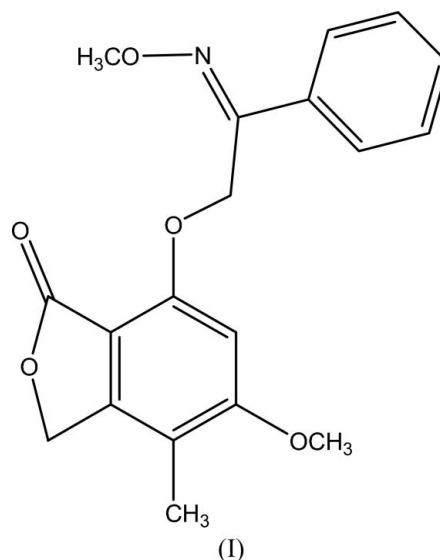
Single-crystal X-ray study  
 $T = 294\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$   
 $R$  factor = 0.045  
 $wR$  factor = 0.144  
Data-to-parameter ratio = 13.3For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(Z)-6-Methoxy-4-(2-methoxyimino-2-phenylethoxy)-7-methylisobenzofuran-3(1H)-one**

The title compound,  $\text{C}_{19}\text{H}_{19}\text{NO}_5$ , was synthesized by the reaction of 6-methoxy-4-(2-oxo-2-phenylethoxy)-7-methylisobenzofuran-3(1H)-one with *o*-methylhydroxylamine hydrochloride in ethanol. It is a new kind of methoxyimino compound containing isobenzofuran. The dihedral angle between the two six-membered aromatic rings is  $104.3(2)^\circ$ .

Received 3 November 2006  
Accepted 15 December 2006

## Comment

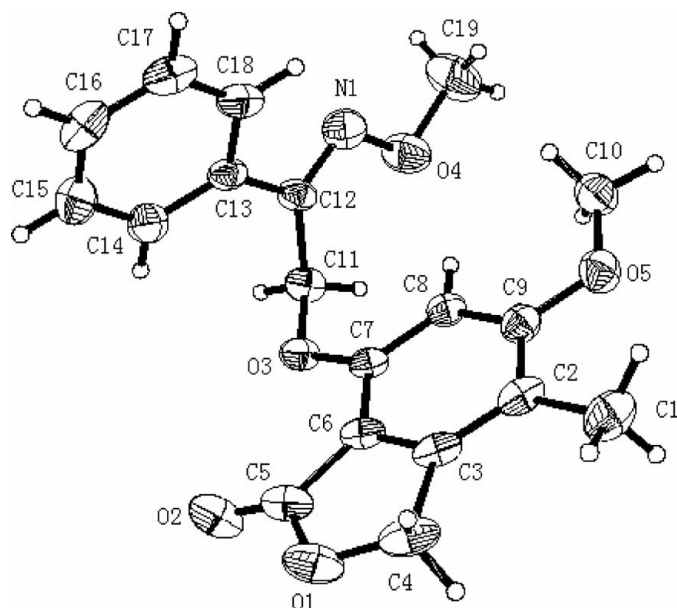
Much attention has been paid to methyloximes containing diaryl groups (Wang *et al.*, 2005). In view of this, we have recently focused on the preparation of methyloximes that contain isobenzofuran. The title compound, (I), was prepared and its structure determined (Fig. 1).



The dihedral angle between the two six-membered aromatic rings is  $104.3(2)^\circ$ . The dihedral angle between the five-membered ring and the fused benzene ring is  $0.7^\circ$ .

## Experimental

The title compound was prepared according to the procedure of Wang *et al.* (2005). To a solution of 1,3-dihydro-6-methoxy-4-(2-oxo-2-phenylethoxy)-7-methyl-3-oxo-isobenzofuran (0.312 g, 1 mmol) in EtOH (20 ml) was added a solution of *o*-methylhydroxylamine hydrochloride (0.17 g, 2 mmol) in EtOH (2 ml). The mixture was heated at reflux for 4 h (TLC monitoring) and evaporated to yield a residual solid. This white solid thus obtained was collected and purified by FC (silica gel, 40/60 ethyl acetate–petroleum ether) to give (I) (0.294 g, 86.2%). Colourless single crystals of (I) were grown by slow evaporation of a chloroform–heptane (3:2) solution.



**Figure 1**  
The molecular structure of (I), with the atom-numbering scheme and 30% probability displacement ellipsoids. H atoms are drawn as spheres of arbitrary radius.

#### Crystal data

$C_{19}H_{19}NO_5$   
 $M_r = 341.35$   
 Monoclinic,  $P2_1/c$   
 $a = 9.3964$  (2) Å  
 $b = 19.445$  (4) Å  
 $c = 9.8521$  (2) Å  
 $\beta = 105.307$  (3)°  
 $V = 1736.3$  (5) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.306$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 294$  (2) K  
 Block, colourless  
 $0.30 \times 0.26 \times 0.20$  mm

#### Data collection

Bruker SMART 1000  
 diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan  
 (SADABS; Sheldrick, 1990)  
 $T_{min} = 0.972$ ,  $T_{max} = 0.981$

8700 measured reflections  
 3057 independent reflections  
 1900 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.033$   
 $\theta_{max} = 25.0^\circ$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.144$   
 $S = 1.02$   
 3057 reflections  
 229 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0775P)^2 + 0.1213P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{max} = 0.001$   
 $\Delta\rho_{max} = 0.36$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.16$  e Å<sup>-3</sup>

All H atoms were positioned geometrically and refined as riding (C–H = 0.93–0.97 Å), with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(methyl C)$ .

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXL97*.

The authors acknowledge support from the Basic Research Project (No. 2002CCA01500) of the MOST (Ministry of Science and Technology of the People's Republic of China).

#### References

- Bruker (1997). *SMART*, *SAINTE* and *SHELXTL*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1990). *SADABS*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Wang, T. C., Lu, P. J., Wong, C. H., Liao, C. H., Tsiao, K. C., Chang, K. M., Chen, Y. L. & Tzeng, C. C. (2005). *Bioorg. Med. Chem.* **13**, 6045–6053.